

SUTYAGIN, A.D.

Climatic characteristics of summer precipitation in the northern
regions of Kirghizia. Trudy Sred.-Az.nauch.-issl.gidrometeor.
inst. no. 8:17-33 '63. (MIRA 17:5)

VLADIMIROV, V.M.; SUTYAGIN, G.N.

Apparatus for determining soil adhesion by means of breaking un-
der field conditions. Trudy Inst. gor. dela Sib. otd. AN SSSR
no.3:319-323 '60. (MIRA 14:4)
(Soil mechanics--Research)

VLADIMIROV, V.M.; SUTYAGIN, G.N.

Apparatus with a built-in recorder for determining the shear
resistance of soil. Trudy Inst. gor. dela Sib. otd. AN SSSR
no.3:324-328 '60. (MIRA 14:4)

(Soil mechanics--Research)

VLADIMIROV , V.M.; SUTYAGIN, G.N.

Laboratory studies of a rotor with inertial unloading. Trudy Inst.
gor. dela Sib. otd. Akad. Nauk SSSR no.7:49-52 '62. (MIRA 16:9)

BARSOV, Nikolay Nikolayevich, dotsent, kand.geograf.nauk; BONIFAT'YEVA, Lidiya Ivanovna, dotsent, kand.geograf.nauk; BURENKO, Sergey Fedorovich, dotsent, kand.geograf.nauk; GITLITS, Semen Aleksandro- vich, dotsent, kand.ekonom.nauk; GUREVICH, Priam Vladimirovich, prof.; DARINSKIY, Anatoliy Viktorovich, dotsent, kand.geograf.nauk; DOLININ, Aleksey Arkad'yevich, dotsent, kand.geograf.nauk; DOROSHKEVICH, Lyudmila Ivanovna, dotsent, kand.geograf.nauk; YEFIMOVA, Yelena Se- menovna, kand.geograf.nauk; LAVROV, Sergey Borisovich, dotsent, kand. geograf.nauk; LEDOVSKIKH, Stepan Ivanovich, dotsent, kand.geograf. nauk; NEVEL'SHTEYN, Grigoriy Solomonovich, dotsent, kand.geograf. nauk; NIKOLAYEVA, Nadezhda Vasil'yevna, dotsent, kand.geograf.nauk; OGANESOV, Vladimir Artem'yevich, kand.geograf.nauk; PINKHENSON, Dmitriy Moiseyevich, dotsent, kand.geograf.nauk; POSPELOVA, Nata- liya Georgiyevna, prof., doktor ekonom.nauk; SEMEVSKIY, Boris Nikola- yevich, prof., doktor geograf.nauk; SUTYAGIN, Pavel Grigor'yevich, dotsent, kand.geograf.nauk; SHTEYN, Viktor Moritsovich, prof., doktor ekonom.nauk; YEROPEYEV, I.A., red.; SMIRNOVA, N.P., red.; TYOTYUNNIK, S.G., red.kart; BORISKINA, V.I., red.kart; KOZLOVSKAYA, M.D., tekhn.red.

[Economic geography of foreign countries; student manual] Ekonomi- cheskaya geografija zarubezhnykh stran; posobie dlia studentov. Moskva, Gos.uchebno-pedagog.izd-vo M-va prosv.RSFSR, 1960. 702 p. # maps (MIRA 13:12)

(Geography, Economic)

TANANAYEV, I.V.; GALKIN, N.P.; SAVCHENKO, G.S.; SUTYAGIN, V.M.

Interaction of UCl_4 and $U(SO_4)_2$ with NaF in aqueous solutions. Zhur.
neorg.khim. 7 no.7:1675-1680, 1962. (MIRA 16.3)
(Uranium chloride) (Uranium sulfate) (Sodium fluoride)

ГОРЬКОЕ, В.

Hundreds of voluntary helpers. MTO 9 no. 9:54-55 8 '63.

(MIRA 17:6)

1. Chlen soveta Obschestvennogo konstruktorskogo byuro Ural'skogo zavoda tyazhelogo mashinostroyeniya imeni Ordzhonikidze.

ACC NR: AP6025662

(A)

SOURCE CODE: UR/0413/66/000/013/0128/0128

INVENTOR: Sutyagin, V. A.

ORG: None

TITLE: Radial-thrust plain bearing with an air cushion. Class 47, No. 183543 [announced by the Izhevsk Machine Building Plant (Izhevskiy mashinostroitel'nyy zavod)]

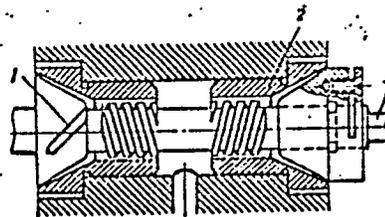
SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 13, 1966, 128

TOPIC TAGS: journal bearing, air lubricated bearing

ABSTRACT: This Author's Certificate introduces a radial-thrust plain bearing with an air cushion which uses a tapered bushing. A shaft with a tapered surface is set in this bushing. The cylindrical part of the shaft is threaded. The carrying capacity of the bearing is increased by forming slots on the tapered support surface. These slots are cut at a negative angle to that of the shaft threads and pass through only a part of the tapered support surface.

SUB CODE: 13/ SUBM DATE: 05Apr65

Card 1/1

1—slots; 2—bushing; 3—
shaft

UDC; 621.822.5

MOROSHKIN, B.F., prof.; KOSTINA, A.A., kand. veter. nauk; IVANSKIY, Ye.F.,
kand. veter. nauk; SUTYAGIN, V.S., kand. veter. nauk

Hemoparasitosis in experimental fascioliasis. Veterinaria 41
no.10:11-12 0 '62. (MIRA 18:11)

СУТЯГИН, В. Я.

Technology

Homemade televisor, Moskva, Gosenergoizdat, 1951.

9. Monthly List of Russian Accessions, Library of Congress, December 1952 ~~1953~~, Uncl.

MR

L

489-L. Porosity of Electrodeposited Nickel. (In Russian.) A. A. Sulaginn and A. P. Vagramian. *Zhurnal Prikladnoi Khimii*, v. 24, Sept. 1951, p. 945-950.

The influence of thickness of deposit, current density, and various surface-active materials was investigated. Some materials were found to increase and others to decrease the number of pores. Up to a limiting value, porosity decreases with increasing thickness. The porosity-current density curve passes through a minimum. Data are charted. 11 ref. (L17)

SUTIYAGIN, V. Ya.

"Amateur Television Receivers," 1952

This book includes a description of how to construct a television receiver using the frequency for channel 1 (49.75 mc for picture and 56.25 for voice). The operation is based on a minimum of tubes and for a distance of not more than 15 or 20 kilometers from the transmitting station. This receiver does not utilize an intermediate frequency. The recommended screen radius for the kinescope circular tube is nine inches.

LXI

SUTYAGIN, V. YA.

SUTYAGIN, V. Ya.; BERG, A. I., redaktor; DZHIGIT, I. S., redaktor; YELIN, O. G., redaktor; KULIKOVSKIY, A. A., redaktor; MOZHICHEVELOV, B. N., redaktor; SMIRNOV, A. D., redaktor; TARASOV, F. I., redaktor; TRAMM, B. F., redaktor; CHECHIK, P. O., redaktor; SHAMSHUR, V. I., redaktor; KRIVOSHEYEV, I. I., redaktor; FRIDKIN, A. M., tekhnicheskiy redaktor.

[Circuits of television receiver scanning devices] Skhemy razvertyvayushchikh ustroystv televizionnykh priemnikov. Moskva, Gos. energ. izd-vo, 1954. 93 p. (Massovaya radiobiblioteka, no. 199) (MLRA 7:9)
(Television--Receivers and reception)

SUTYAGIN, V. Ya.

KULIKOVSKIY, A.A.; SUTYAGIN, V.Ya., redaktor; FRIDKIN, A.M., tekhnicheskiy
redaktor

[New developments in amateur radio receivers] Novoe v tekhnike
liubitel'skogo radiopriema. 2-e izd., perer. Moskva, Gos. energeti-
cheskoe izd-vo, 1954. 174 p. (Massovaia radiobiblioteka, no.207)
[Microfilm] (MIRA 8:3)
(Radio-Receivers and reception)

9(2)

SOV/112-59-3-5938

Translation from: Referativnyy zhurnal. Elektrotehnika, 1959, Nr 3, p 241 (USSR)

AUTHOR: Sutyagin, V. Ya.

TITLE: Temperature-Compensated Transistorized Power Amplifiers
(Usiliteli moshchnosti na kristallicheskikh triodakh s termokompensatsiyey)

PERIODICAL: V sb.: Poluprovodnikovyye pribory i ikh primeneniye. Nr. 1, M.,
"Sov. radio," 1956, pp 354-401

ABSTRACT: Properties of common-emitter circuits are considered whose parameters are: $G_{11} = Y_{be} + Y_{bk} \approx Y_{be}$; $G_{21} = S - Y_{bk} \approx S$; $G_{12} = -Y_{bk}$; $G_{22} = Y_{ke} - Y_{bk} \approx Y_{ke}$ with $Y_{bk} \ll Y_{be}$; $Y_{bk} \ll Y_{ke}$; $Y_{bk} \ll S$. Plots are presented of transistor parameters against the voltages U_k and U_b , as well as a plot of the power amplification factor against U_b . Simple and push-pull output stages are designed graphically. For a simple stage such value of U_{b_0} should be selected which corresponds to not very curved static

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SOV/112-59-3-5938

Temperature-Compensated Transistorized Power Amplifiers

characteristic $i_k(u_b)$ with $u_k = U_{ko}$. After U_{mvkh} and the operating point are selected on the basis of the static characteristic i_k, u_k family, the dynamic characteristic should be plotted, R_n determined, and P_{vykh} calculated. Then, the nonlinear-distortion factor k_f, I_{sr} , and the stage efficiency $\eta = P_{vykh}/I_{sr}U_{ko}$ should be computed. The resistance R_{vkh} is determined from the dynamic characteristic $i_b(u_b): R_{vkh} = \text{ctg } \varphi$, where φ is the angle of slope of a tangent to the dynamic characteristic at the operating point. The signal power consumed by the transistor is $P_{vkh} = U_{mvkh}^2/2R_{vkh}$. The class AB operation of the push-pull stage can also be designed graphically. First, such a maximum base voltage value U_{bmaks} is selected that no serious non-linear distortion and no large losses in the base circuit would occur. Having selected U_{ko} so that $P_{k maks} \ll P_{k dop}$, a dynamic characteristic on the basis of $i_k(u_k)$ family is plotted and the operating point U_{bo} is selected. The

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Temperature-Compensated Transistorized Power Amplifiers

oscillatory power is $P_k = I_{smaks} (U_{ko} - U_{kmin})/2 \approx I_{smaks} U_{ko}/2$ because $U_{kmin} = 0.3 - 0.5$ v; the load resistance to the total current is $r_{ks} = U_{ko}/I_{smaks}$; the input power is $P_{vkh} = P_b + P_{sh} = S_b [U_{mb}/(1 + S_b R_g)]^2 + 2U_{mb}/R_{sh}$, where S_b is the transconductance of the perfect characteristic ($S_b = 1/R_b$); R_g is the R_{vykh} resistance of a preceding stage in terms of the input-transformer secondary and allowing for R_{sh} . To reduce nonlinear distortion, the resistance should be assigned $R_{sh} \ll R_{vkh}$ at U_{bmaks} . Then, the dynamic characteristic for the component currents $i_k' - i_k'' = i_s = f(u_{b\sim})$ should be plotted, and K_f and η should be found. To design a power amplifier for functioning within a wide temperature range, it is necessary to allow for transistor-parameter variations which almost follow a linear law. Having designed the amplifier on the basis of a curve family corresponding to the room temperature, the temperature-compensating components should be designed on

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SOV/112-59-3-5938

Temperature-Compensated Transistorized Power Amplifiers

the basis of static characteristics $i_k = f(u_b)$ with $U_{k0} = \text{const}$ for various t° .

The latter characteristics can be obtained from the room-temperature static characteristics by applying an empirical formula

$$i_{k\Delta t} = \Delta t / 3 \sqrt[3]{i_k u_b S - 2I_k}$$

where Δt is the temperature rise over the room temperature. An R_e , S_e chain should be connected to the emitter circuit for purposes of temperature compensation. The output power can be maintained $P_{vykh} = \text{const}$ with changing input voltage U_{tb} that can be computed for the maximum positive temperature. U_{tb} can be adjusted by electromechanical means or a thermistor-type divider can be used at the circuit input. Examples of graphical design of temperature-compensated amplifiers are cited

V.I.A.

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SUTYAGIN, V. Ya.

9(4) p 3

PHASE I BOOK EXPLOITATION SOV/1778

Nauchno-tekhnicheskoye obshchestvo priborostroitel'noy promyshlennosti. Moskovskoye pravleniye

Tranzistornaya elektronika v priborostroyeni; sbornik trudov konferentsii (Transistor Electronics in the Instrument-making Industry; Collection of Conference Transactions) Moscow, Oborongiz, 1959. 289 p. 1,400 copies printed.

Ed.: N.I. Chistyakov, Doctor of Technical Sciences, Professor; Ed. of Publishing House: S.D. Khametova; Tech Ed.: V.P. Rozhin; Managing Ed.: A.S. Zaymovskaya, Engineer.

PURPOSE: The book is intended for scientific and engineering personnel of the instrument-making and radio industries engaged in the development of electronic and radio equipment.

COVERAGE: The authors of this collection of articles discuss the theory, principle of operation, calculation and application of electronic circuits using transistors. They also

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Transistor Electronics (Cont.)

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feedback. Operation of circuits used for experimentally determining transistor parameters is also discussed. There are 4 references of which 2 are Soviet, and 2 English

N.K. Povarov, Candidate of Technical Sciences. Electronic Devices Fed by Current Generators 25

The author describes the static and dynamic characteristics of nonlinear elements and discusses their equivalent circuits. He also describes the operation and characteristics of vacuum phototubes, vacuum-tube amplifiers, transistors, cascade amplifiers, and oscillators connected to a current generator. There are 8 references of which 7 are Soviet and 1 English.

V. Ya. Sutyagin, Engineer. Average-current Transistor Amplifiers 39

The author discusses the operation and characteristics
Card 3/12

05212

SOV/142-2-3-20/27

9(2,3)
AUTHORS:

Kulikovskiy, A.A., Potryasay, V.F., Sutyagin, V.Ya., Ryzhkov, A.S.

TITLE:

The Terminology in the Field of Transistor Electronics

PERIODICAL:

Izvestiya vysshikh uchebnykh zavedeniy, Radiotekhnika, 1959, Vol 2, Nr 3, p 378 (USSR)

ABSTRACT:

The authors refer to the article by T.M. Agakhanyan, B.N. Kononov and I.P. Stepanenko, titled "The Terminology in the Field of Transistor Electronics", published in Izvestiya vysshikh uchebnykh zavedeniy, Radiotekhnika, 1958, Vol 1, Nr 4. The authors agree in principle with the suggestions made in the aforementioned article and present some of their own ideas as an addition. For example the Russian terms "baza" (base) and "tranzistor" (transistor) should be sanctioned, although there might be some conflict with the term "poluprovodnikovyy diod" (semiconductor diode) which also belongs to the transistor class. The authors regard the terms "dyrochnyy tranzistor" ('hole' transistor) and "elektronnyy tranzistor" ('electron' transistor) as superflous and recommend the designation p-n-p or n-p-n transistor. Similar suggestions were made for the classification of diode types.

Card 1/2

9,4310 (1139, 1143, 1159)

30500
S/194/61/000/008/049/092
D201/D304

AUTHORS: Potryasay, V.F., Ryzhov, A.S. and Sutyagin, V.Ya.

TITLE: Transistor noise

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika, no. 8, 1961, 14, abstract 8 D98 (V sb. Poluprovodnik. pribory i ikh primeneniye, v. 5, M., Sov. radio, 1960, 107-158)

TEXT: It is shown that the noise factor F should be taken as the basic factor determining the noise properties of transistor amplifiers. Possibilities are considered of replacing the transistor noise by a noisy four-pole of the equivalent transistor circuit, having internal noise sources. The bloc-diagram is given for measuring the noise parameters. Formulae are given for evaluating F in the region of white noise and at HF from the equivalent circuit of Giacoletto, containing 4 uncorrelated noise sources corresponding to the following: Thermal noise of the ohmic base resistance; shot

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SUTYAGIN, V.S.

USSR/Zooparasitology - Tics and Insects (Disease Transmitters) P-3

Abs Jour : Referat Zhur - Biologii, No 16, 1957, 70194

Author : Dudkina, M.S., Sutyagin, V.S.Inst : Sb. nauch. L'vovsk. n.i. In-ta epidemiol. mikrobiol.
i gigeny. L'vovsk. Un-t 1956, 63-71Title : Ecology *An. bifurcatus* in L'vov Region

Orig Pub : See Inst.

Abstract : In the Lvov region, the *Anopheles bifurcatus* is widely distributed in the habitations of the raised hilly region where it constitutes by day count 13-16% of the total number of mosquitoes. Hatching places - spring and sub-soil fed, temporary rain puddles with a T. of 17-23 deg C. Larvae of all ages hibernate (mainly III) in unfrozen swampy areas of streams and fish-ponds. In the course of a season there are 3-4 generations; there is no decrease in the summer.

In cool, moist years, the large number of mosquitoes is

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- 34 -

GRIGOROV, O.N.; SUTYAGIN, Ye.I.

Analytical method for the determination of the number of transfer through membranes. Vest. LGU 18 no.22:163-165 '63.

Transfer number of ions in aluminum oxide films as dependent on current direction and density. ~~Ibid.:163-167~~
(MIRA 17:1)

GREKMOV, O.S.; SUYANIN, Ye.I.

Effect of concentration changes in membranes on the velocity
of electroosmotic transport. Vest. ION ID no.16:101-109 '64.
(MIRA 17:11)

SIDOROVA, M.F.; BUTYAGIN, Ye. I.

Influence of concentration polarization upon the electrochemical
activity of membranes. Vest. LGU 19 no.22 82-87 '64
(MIRA 18:1)

SUTYAGINA, A. A.

USSR/Chemistry - Electrochemistry, May/June 52
Nickel

The Effect of Alternating Current on the Electrodeposition of Nickel," A. F. Vagrameyan, A. A. Sutyagina, Inst of Phys Chem, Acad Sci USSR

"Iz Ak Nauk, Otdel Khim Nauk" No 3, pp 410-414

On application of above certain current densities, the structure of the deposit becomes macrocryst. The number of pores in the electrolytic Ni deposit decreases. With increased frequency of the ac, its action gradually ceases. A possible mechanism of

22075

the action of ac on the change of the structure of the deposit and on overvoltage in discharge of Ni ions is suggested.

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SUTYACINA, A.A.

SUTYACINA, A.A. "The Effect of Alternating Current on the Processes of Electrocrystallization of Certain Metals." Acad Sci USSR. Inst. of Physical Chemistry. Moscow, 1956. (Dissertation for the Degree of Candidate in Chemical Science)

So: Knizhnaya Letopis', No. 18, 1956,

S/564/57/000/000/003/029
D258/D307

AUTHORS: Gorbunova, K. M., Popova, O. S., Sutyragina, A.
A., and Polukarov, Yu. M.

TITLE: Mechanism of growth and structure of metallic
deposits formed during electrocrystallization

SOURCE: Rost kristallov; doklady na Pervom soveshchanii
po rostu kristallov, 1956 g. Moscow, Izd-vo
AN SSSR, 1957, 58-66

TEXT: The present article is a review of some of the authors' earlier studies and other work; attention is focused on some regularities concerning the cathode deposition of metals, as dense or porous, dendritic, coatings. An increase in the electrolysis current leads to an increase in the surface of the crystallizing metal, leading to an increase in the number of crystals or to dendritization, according to conditions. Powder deposition (fine dendrites) occurs at the saturation current.

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137-58-4-7842

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 4, p 211 (USSR)

AUTHORS: Gorbunova, K. M., Popova, O. S., Sutyagina, A. A., Polukarov, Yu. M.

TITLE: Mechanism of Growth and Structure of Precipitates of Metals Produced by Electrical Crystallization (Mekhanizm rosta i stroyeniye osadkov metalla, vznikayushchikh pri elektrokristallizatsii)

PERIODICAL: V sb.: Rost kristallov. Moscow, AN SSSR, 1957, pp 58-66

ABSTRACT: Certain principles of the growth on the cathode of an electrolytic cell of deposits (D) of metal in the form of dense coatings or loose dendritic structures are examined. K. M. Gorbunova shows that when single crystals are formed, an increase in current I results in the $I/\Sigma S-K$ ratio remaining constant because of the increase in the surface of growth. This latter results in a transition from growth of the single crystal to the growth of multicrystalline D (a relatively high concentration of discharging ions occurs) or to a growth of dendritic D (a low concentration of discharging ions). Dense polycrystalline D grow when $\Sigma S-S$ is attained at the cathode. Subsequently, further increase in I can occur only when there is

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137-58-4-7842

Mechanism of Growth and Structure (cont.)

a drop in the concentration of ions at the cathode. At a given I , the ion concentration at the cathode may prove to be close to zero: the maximum diffusion current is attained (MDC). Powder D form upon electrocrystallization under MDC conditions. The particles of the powder D are extremely fine dendrites, the angles between the branches of which are determined by the crystallographic nature of the metal. For Zn powder, the angle is 60° . In dense crystalline D the anisotropy of properties such as the magnetic, the linear compressibility, resistivity, thermal expansion, resistance to corrosion, etc., are determined by texture (orientation of all the crystals of the D in a given crystallographic direction). The authors hold that in the case of D with crystals above a certain size and small internal stresses (IS), it is more accurate to regard texture as "growth texture." Texture comes into being as the result of competition between crystals of different orientations, as the ionic building blocks brought up to the growing crystals are put to use. The change in the texture axis with change in the conditions of electrolysis is explained by the change in the ratios of the growth rates in different directions. The unique adherence of the texture of Zn and Cd D to a 6th-order axis $[0001]_c$ on application of an alternating current, with the surfaces bounded not by apices but by the faces of the base, may be explained in terms of the concepts developed by Kaishev and Bliznakov. X-ray and magnetic studies have made it possible to determine

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137-58-4-7842

Mechanism of Growth and Structure (cont.)

that IS anisotropy exists in Ni deposits, and also that the IS of Ni is not directly related to the amount of occluded H₂. Introduction of brightening agents in the bath leads to the formation of deposits not having the definite, clearly defined boundaries characteristic of crystals, and the D consist of rounded forms.

O. P.

1. Cathodes--Deposits--Structural analysis
2. Metals--Crystallization--Structural analysis

Card 3/3

32-1-19/55

AUTHOR: Sutiyagina, A.A.

TITLE: The Determination of Sulphur Impurities in Galvanic Precipitations of Nickel and Copper by the Radiochemical Method (Opredeleniye vklyucheniye sery v gal'vanicheskikh osadkakh nikelya i medi radiokhimicheskim metodom).

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 1, pp. 43-44 (USSR)

ABSTRACT: In galvanoplastics organic compounds with a sulphur content are used for processes of precipitation. The metallic precipitations in this case contain the sulphur impurities [Ref. 1-4]. In order to determine them the method mentioned here is recommended as well as for the investigation of the galvanic precipitate of nickel (in the nickel bath) or copper (in the copper bath). As an indicator thiourea with a content of radioactive sulphur (S^{35}) is used. The nickel bath usually contains: 160 g/l $NiSO_4 \cdot 7H_2O$, 30 g/l H_3BO_3 and 10 g/l NaCl; the copper bath: 250 g/l $CuSO_4 \cdot 5H_2O$ and 10 g/l H_2SO_4 . The thiourea content in the first case amounts to 0,1-0,4 g/l, and in the second case to 0,04-0,2 g/l. The activity of the initial solution is judged according to the "dry" sample,

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The Determination of Sulphur Impurities in Galvanic Precipitations of Nickel and Copper by the Radio-chemical Method

32-1-19/55

(which is obtained by applying the solution on to a special plate and by vaporization. If the content of thiourea in the initial solution is known, and after the activity of this solution has been determined according to the "dry" sample, the content of sulphur in the metallic precipitate can be determined. This method was tested by chemical analysis and only slightly lower results were obtained which can be explained by the different manner of preparing the samples. There are 2 tables, and 6 references, 1 of which is Slavic.

ASSOCIATION: Institute for Physical Chemistry AN USSR (Institut fizicheskoy khimii Akademii nauk SSSR).

AVAILABLE: Library of Congress

Card 2/2 1. Sulfur-Impurities-Determination 2. Chemical analysis

76-32-4-9/43

AUTHORS: Gorbunova, K. M., Sutyagina, A. A.

TITLE: The Gloss of Electrolytic Deposits and its Connection With the Internal and External Structure of the Deposit (K voprosu o bleske elektroliticheskikh osadkov i svyazi yego s vnutrennim i vneshnim stroeniyem osadka)

PERIODICAL: Zhurnal Fizicheskoy Khimii, 1958, Vol. 32, Nr 4, pp. 785 - 790 (USSR)

ABSTRACT: By changing the conditions of deposition an essential change of the crystal lattices is caused in many metals; this is made reason for the present paper to deal with the problem of the influence of geometric factors or the surface relief respectively on the gloss. It is assumed that the gloss is determined by the size of the crystals forming the deposit, and that they only shine when the crystal dimensions are sufficiently small. Similar statements were made by Wittum (Reference 1) and S. M. Kochergin (Reference 3,2) as well as by A. T. Vagramyan and Z. A. Solov'yeva (Reference 4,5). Wood (Reference 6) was the first to point out the connection between the gloss and the crystal orientation, while S. P. Makar'yeva (Reference 7)

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76-32-4-9/43

The Gloss of Electrolytic Deposits and its Connection With the Internal and External Structure of the Deposit

observed changes of the texture, just as well as V. I. Arkharov (Reference 8), L. S. Palatnik (Reference 9), V. I. Arkharov (Reference 8), G. F. Kosolapov and B. Yu. Mett (Reference 10), N. T. Kudryavtsev and B. V. Ershler (Reference 11) investigated electrodeposits of zinc. In the last years the opinion became popular which originated from K. M. Gorbunova et al. (Reference 12,13) that gloss is not clearly determined by the dimension and the character of mutual orientation of the crystals. This was also noticed in a similar way by Smith, Keeler and Read (Reference 14) as well as by Read and Weil (Reference 15) and Klark and Simonsen (Reference 16). In the present work the new data obtained by A. A. Sutyagina (Reference 19) are investigated and the influence of a.c. on the electrocrystallization of a number of metals is dealt with. As can be seen from the results and the mentioned figures the crystal dimension is no clear criterion for the gloss; some examples are given for this. A connection between the texture

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76-32-4-9/43

The Gloss of Electrolytic Deposits and its Connection With the Internal and External Structure of the Deposit

and the gloss was also not observed, and the assumptions by K. M. Gorbunova, T. V. Ivanovskaya and N. A. Shishakov (Reference 12) were proved. The electrochemical investigations of glossy and dead zinc depositions showed, however, that the increase of surface polish and the absence of sharp raisings increase the gloss so that a connection between the reflection coefficient and the surface relief is assumed. There are 8 figures, 1 table and 20 references, 13 of which are Soviet.

ASSOCIATION: Akademiya nauk SSSR, Institut fizicheskoy khimii, Moskva (Moscow Physico-Chemical Institute, AS USSR)

SUBMITTED: November 26, 1956

AVAILABLE: Library of Congress

Card 3/3 1. Metals--Electrodeposition 2. Crystals--Lattices 3. Metals --Surface properties

82577

S/081/60/000/006/001/008
A006/A001

Some Regularities of Metal Electrocrystallization Under the Effect of A.C.

believe that in the case of Zn and Cd, the structural changes are connected with the surface activation of the anode current component as a result of the potential shift toward the anode zone; in the case of Ni, changes in the structure do not depend on the nature of surface-active additions and are caused by the difficult formation of Ni hydroxide in the layer near the electrode, due to H ionization and the accumulation of H+ ions during the anode pulses. ✓

Z. Solov'yeva

Translator's note: This is the full translation of the original Russian abstract.

Card 2/2

SOV/76-33-9-15/37

Electrocrystallization of Nickel by the Application of Alternating Current.
I. Structure of Deposits From Electrolytes Not Containing Additions of Surface Active Substances

= 2 a/dm², frequency 50 cycles) differ but very little from those obtained with the direct current. With an increase in density of a.c. to 7.5 a/dm², however, the nickel deposit crystals grow to a great extent, and so does the change in texture (Figs 1-7), while reflection is diminished. In alternating current frequencies of 500 cycles the last mentioned changes are not observable. A strong change of the nickel deposit occurs only with the ratio: density of a.c./density of d.c. > 1 and with lower frequencies of the alternating current, in which connection the acidity of the electrolyte plays an important part. Similar observations were made with the electrolyte 160 g/l NiSO₄·7H₂O, 45 g/l NiCl₂·6H₂O, 30 g/l H₃BO₃. Investigations were extended to the change in potential of nickel separation under the abovementioned conditions with the aid of a special apparatus (Fig 8), and a loop oscillograph. By the application of alternating current (50 cycles) the cathode potential changes according to a sinusoid, the axis of which shifts to more positive values (Table), as compared to the potential of nickel separation in the application of a direct

Card 2/3

SOV/76-33-9-15/37

Electrocrystallization of Nickel by the Application of Alternating Current.
I. Structure of Deposits From Electrolytes Not Containing Additions of Surface
Active Substances

current only. (Fig 9, oscillograms). There are 9 figures,
1 table, and 10 references, 7 of which are Soviet.

SUBMITTED: February 22, 1958

Card 3/3

5(4), 18(7)
AUTHORS:

Sutyagina, A. A., Gorbunova, K. M.

05804
SOV/76-33-10-2/45

TITLE:

Electrocrystallization of Nickel Under Alternating Current Conditions. II. Electrolytes With Additions of Surface-active Substances

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 10, pp 2128-2134 (USSR)

ABSTRACT:

In continuation of a previous paper the effect exerted by surface-active substances on the electrocrystallization of nickel is investigated. The method already described (Refs 1-4) is employed. The alternating current electrodeposition of nickel was investigated in the presence of one of the following substances: glass-producing additions of the anionic type of 2,6 - 2,7-naphthalene disulfonic acid; additions the effect of which is based on the separation of sulfur ions in the solution (thiourea, allylthiourea); additions of the molecular type (formaldehyde); additions of the cationic type (aniline). Nickel deposits which were obtained at direct current ($D_G = 2 \text{ a/dm}^2$) from electrolytes with one of the above-mentioned substances have a lustrous surface on which no crystal faces

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SOV/76-33-10-2/45

Electrocrystallization of Nickel Under Alternating Current Conditions.
II. Electrolytes With Additions of Surface-active Substances

were observed, not even with an EM-3 electron microscope (Fig 1). The radiographs of these deposits have blurred lines (Fig 2). The calculations of the radiographs carried out according to A. A. Sutyagina's method (Ref 4) showed that in the lustrous nickel deposits an X-ray diffraction pattern existed along the (001) axis. In accordance with data by K. M. Gorbunova, T. V. Ivanovskaya and N. A. Shishakov (Ref 5), an image with diffraction circles was obtained on electronograms (Fig 3) of nickel deposits resulting from solutions with an addition of 2,6 - 2,7-naphthalene disulfonic acid. At a ratio of $D_{a-c} : D_{d-c} > 1$ (alternating current of 50 cycles) a strong effect on the structure of the nickel deposit was observed; irrespective of the type of additions, in most cases nickel deposits were obtained with an X-ray diffraction pattern along the (001) axis. Nickel deposits obtained at direct current ($D_G = 2 \text{ a/dm}^2$) from electrolytes with additions of formaldehyde or aniline possessed weak reflexion (Table), whereas nickel deposits from electrolytes with allylthiourea obtained at $D_G = 1 - 2 \text{ a/dm}^2$ reflect strongly (Table). An increase of the

Card 2/3

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SOV/76-33-10-2/45

Electrocrystallization of Nickel Under Alternating Current Conditions.
II. Electrolytes With Additions of Surface-active Substances

alternating-current frequency reduces its effect so that at frequencies of 5,000 cycles the nickel deposits show the same structure as those obtained at direct current only. The experimental results are explained and the papers by A. A. Rotinyan, V. Ya. Zel'des (Ref 10), I. I. Zhukov, Z. D. Pigareva (Ref 11), N. A. Izgaryshev, S. Berkman (Ref 16) and Yu. M. Polukarov (Ref 17) are mentioned; a change of the adsorption conditions on the cathode is taken into consideration (at a potential shift) as well as a change of the H^+ -ionic concentration on the cathode (caused by a periodic change of the cathodic and anodic process). There are 8 figures, 1 table, and 17 references, 14 of which are Soviet.

ASSOCIATION: Akademiya nauk SSSR, Institut fizicheskoy khimii
(Academy of Sciences of the USSR, Institute of Physical Chemistry)

SUBMITTED: February 22, 1958

Card 3/3

On the Problem of the Mechanism of Sulfur Inclusion in Electrodeposits

68822

S/020/60/131/01/037/060
B004/B011

changes. The determination of the sulfide content in the nickel deposits showed that most of the sulfur was present as sulfide, with the stoichiometric ratio not corresponding to any known sulfide. Such a formation of sulfur-enriched sulfides, whose additional sulfur was bound but loosely, had already been mentioned in reference 7. On the other hand, but little sulfide sulfur was found in the case of zinc. The largest portion of sulfur is present in the electrolytic zinc deposit in the form of unchangedly adsorbed molecules of the added substance. The particular behavior of nickel is explained by the circumstance that freshly electrodeposited nickel contains hydrogen, and therefore approaches Raney nickel as to its properties, the latter being known to be capable of separating sulfur from organic and inorganic compounds in a quantitative manner (Ref 9). There are 1 table and 9 references, 8 of which are Soviet.

ASSOCIATION:

Institut fizicheskoy khimii Akademii nauk SSSR (Institute of Physical Chemistry of the Academy of Sciences, USSR)

PRESENTED:

November 10, 1959 by A. N. Frumkin, Academician

SUBMITTED:

October 5, 1959

Card 2/2

SUTYAGINA, A.A.; GORBUNOVA, K.N. (Moskva)

Electrocrystallization of some metals in the presence of the additions of sulfur-containing surface active agents.

Part 1: Effect of the electrolysis conditions on the amount of impurities in zinc deposits, and the mechanism of their incorporation. Zhur.fiz.khim. 35 no.8:1769-1773 Ag '61.

(MIRA 14:8)

1. Institut fizicheskoy khimii AN SSSR.
(Zinc-Plating)
(Sulfur organic compounds)

5.1310 1087 1521

29985

S/076/61/035/011/006/013
B110/B147

AUTHORS: Sutyagina, A. A., and Gorbunova, K. M. (Moscow)

TITLE: Study of the electrocrystallization process of some metals in the presence of sulfur-containing surface-active admixtures. II. Effect of the conditions of electrolysis on the amounts of impurities in nickel deposits and the properties determined by the above conditions

PERIODICAL: Zhurnal fizicheskoy khimii, v. 35, no. 11, 1961, 2514-2523

TEXT: The dependence of the composition of Ni deposits on the type and concentration of admixtures, the acidity of the solution, and the current density at the cathode were studied to elucidate the nature of electrodic processes leading to the formation of indistinctly crystalline deposits. A solution of 160 g/liter of $\text{NiSO}_4 \cdot 7\text{H}_2\text{O}$, 30 g/liter of H_3BO_3 , and 10 g/liter of NaCl with different pH values was used as electrolyte. Thiourea, allyl thiourea, β -sulfonaphthalic acid, 2,6- 2,7-disulfonaphthalic acid, α -naphthol disulfonic acid, α -naphthol sulfonic acid, sodium thiosulfate, thiosalicylic acid, diphenyl thiourea, and potassium thiocyanate were used as admixtures. The amount of occluded sulfur increases with increasing

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Study of the electrocrystallization ...

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S/076/61/035/011/006/013
B110/B147

concentration of the admixture in acid electrolytes yielding larger amounts of occluded sulfur. The sulfur content is decreased by an increase of current density. The sulfur content is less when admixtures containing sulfur on a benzene ring or bound to oxygen are used. Since the overall-sulfur-to-carbon ratio is higher than in the organic admixtures, chemical conversions are assumed to take place in the deposit. Irrespective of the structure of the admixture, $\geq 90\%$ of S is bound to Ni as sulfide. The higher carbon content probably caused by adsorption of the conversion products of the initial admixture is explained as being due to the removal of S from organic compounds by freshly deposited, H_2 -saturated Ni (similar to Raney nickel). A temperature increase leads to an increase of the S content, despite of the desorption of surface-active agents. S-Ni ratio in the deposit was several times found to be equal to 2.5. The constant ratio found after extraction with CCl_4 and acetone indicates a loose chemical bond since the S-Ni ratio falls to 0.5-0.7 on heating to $130^\circ C$. X-ray analysis revealed a structural change at $400-450^\circ C$: After a reaction time of 1.5 hrs in the solid state, Ni and Ni_3S_2 were formed from the nickel and the high-sulfur primary sulfide phases under exothermic conditions (as has been thermographically established). Heat absorption at $600^\circ C$ is due to

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S/076/61/035/011/006/013

B110/B147

Study of the electrocrystallization ...

the melting of a eutectic mixture consisting of Ni and the decomposition products of Ni_3S_2 . The melting point of Ni is lowered by the occlusion of sulfur. The reflection coefficient of Ni deposits obtained from low-acidity electrolytes is lowered. Gloss is, however, not directly dependent on the sulfur content. The reflection coefficient is also raised when the current density is increased. The hardness of an Ni deposit obtained at 1.4 a/dm^2 (pH = 5.5), which was measured with a PNT-3 (PMT-3) device was about 210 kg/mm^2 , that of an Ni deposit containing 0.69% S, prepared with 0.1 g/liter of thiourea (TU), was about 570 kg/mm^2 . With increasing current density (2.6 a/dm^2), hardness at 0.1 g/liter of TU and at a sulfur content of about 0.42% decreases to 456 kg/mm^2 , while it is increased in the absence of admixture (about 310 kg/mm^2 at 3.3 a/dm^2). At a sulfur content of about 5%, a change in the lattice constant of Ni by about 0.04% occur. Deposits from electrolytes with pH values of 2.5 to 2.6 and low current densities as well as from electrolytes with pH values of 5.5 and with TU additions of 0.3 to 0.5 g/liter showed no texture. Deposits obtained at high current densities from electrolytes with a pH value of about 5.5 and containing 0.1 to 0.2 g/liter of TU as well as from higher-acidity electrolytes showed textures oriented parallel to [001]. In the Card 3/4

SUTYAGINA, A.A.; GORBUNOVA, K.M.; GLAZUNOV, M.P.

Mechanism of chemical nickel plating reaction. Dokl. AN SSSR
147 no.5:1133-1136 D '62. (MIRA 16:2)

1. Institut fizicheskoy khimii AN SSSR. Predstavleno akademikom
V.I. Spitsynym.
(Nickel plating) (Reduction, Chemical)

SUTYAGINA, A.A.; GORBUNOVA, K.M.; GLAZUNOV, M.P.

Reaction mechanism of nickel reduction by hypophosphite studied with the aid of deuterium as tracer. Part 1: Solutions without organic additives. Zhur. fiz. khim. 37 no.9:2022-2026 S '63.

(MIRA 16:12)

1. Institut fizicheskoy khimii AN SSSR.

SUTYAGINA, A.A.; GORBUNOVA, K.M.; GLAZUNOV, M.P.

Study of the mechanism of nickel reduction by hypophosphite with the use of deuterium as tracer. Part 2: Reduction of nickel in hypophosphite solutions with organic acid salts added. Zhur.fiz.khim. 37 no.10:2214-2219 0 '63.
(MIRA 17:2)

1. Institut fizicheskoy khimii AN SSSR.

SUTYAGINA, A.A.; GORBUNOVA, K.M.

Mechanism of phosphorus inclusion in the chemical reduction of
some metals by hypophosphite. Zhur. prikl. khim. 37 no.8:1676-
1681 Ag '64. (MIRA 17:11)

VAGRAMYAN, A. T.: SUTYAGINA, A. G.

Nickelplating

Effect of the alternating current on electrodesposition of nickel.
Izv. AN SSSR Otd. Khim. nauk No. 3, 1952

9. Monthly List of Russian Accessions, Library of Congress, November 195~~2~~, Uncl.

Chronicle. All-Union Competition for the Best Students-paper Concerning Chemistry and Chemical Technology for the Scholastic Year 1957-1958 301/513-2-2-51/31

technologically Institut legkovy proyshlennosti (Moscow technological institute for Light Industry) V. N. Gorodilov; study of the cathodical Polarization at the Precipitation of Chloride from Sulphide-solutions- by the Fifth-year students of the Faculty of Politechnicheskii Institut (Ural Politechnical Institute) V. G. Petrovavitskiy; "Gold Extractions from Mercury Cyanide-solutions" by the Fifth-year students of the Faculty of Metallurgical Institute (Moscow Metallurgical Institute) D. I. Mendeleev; A. V. Gokhlov; "Chemical-technological Institute (Leningrad) V. A. Sorzavov, and M. Brak; "Some Investigations of the Toluene- α -Naphthol Containing Carboxyl" by the Fourth-year-students of the Faculty of Politechnicheskii Institut (Leningrad) G. I. Kosarova and T. A. Shadrinova; "Investigation of the Cathodic and Anodic Processes at Gold-plating" by the Fifth-year-student of the Leningradskiy tekhnicheskii Institut in Leningrad (Leningrad technological institute in Leningrad) E. A. Rozova; "Spectral Determination of Polymers and Tungsten in Tri-hetero-polymide" by the third-year-student of the Kishinevskiy gosudarstvennyy universitet (Kishinev State University) V. I. Degarev; "Research on the Oxidation of Benzoin-ethane by Benzoin-fat in Pome-condition" by the Fourth-year-student of the Kazanaki khimiko-tekhnicheskii Institut (Kazan' Chemical-technological Institute) L. I. Bakhina; "Investigation of the Oxidation of Benzoin-fat" collectively, the competition has been held in the cities of Leningrad, Moscow, Makhachkala, and other cities. The high standard of the scientific research work in the circles of the Studentskoye Nauchnoye Obshchestvo (Students-Scientific Societies) of many universities.

CARD 4/5

10

L 24857-66 EWI(m)/EWP(v)/EWP(j)/T WW/RM

ACC NR: AP6009441 (A)

SOURCE CODE: UR/0377/65/000/003/0041/0048

AUTHORS: Umarov, G. Ya. (Candidate of physico-mathematical sciences); Vil'kova, S. N.; Ayzenshtat, Ye. L.; Novikova, I. A.; Sutyagina, V. M.

ORG: Physicotechnical Institute, AN UzSSR (Fiziko-tekhnicheskiy institut AN UzSSR) 61 2

TITLE: Producing aluminum mirrors on asbestos cement by the conversion method

SOURCE: Geliotekhnika, no. 3, 1965, 41-48

TOPIC TAGS: solar energy conversion, metal plating, asbestos product, aluminum, epoxy plastic, resin, light reflection coefficient/ ED-5 resin

ABSTRACT: The use of low-cost asbestos cement as the body of solar concentrators is described. Epoxy resin ED-5 is used to create a smooth surface on one side of the cement for metallization. This resin shows a small shrinkage as compared with other materials. The resin (15-20 g) with 8% hardener was applied to a 12 x 6-cm plate of asbestos cement and was pressed with a steel beam weighing 3 kg. It was shown that an optically accurate mirror surface can be created by the conversion method (see Fig. 1). A study of the mirror layer showed that its adhesion.

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L 24857-66

AGC NR: AP6009441



Fig. 1. Reflection factor before and after (dotted and continuous curves) conversion versus amount of faience flour (a), graphite (b), and sawdust (c) added to resin.

exceeded by a factor of 5--6 the adhesion of a mirror surface produced by vaporization in a vacuum. Causes of fogging of the reflecting surface with time are explained, and methods of their elimination are shown. Orig. art. has: 5 photographs, 1 graph, and 1 table.

SUB CODE: 10, 20/ SUBM DATE: 23Apr65/ ORIG REF: 002

2/2 dda

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871.6
SOV/126-7-3-31/44AUTHORS: Nakhutin, I. Ye. and Sutyagina, Ye. I.TITLE: Absorption of Hydrogen by Palladium at Low Temperatures
(Pogloshcheniye vodoroda palladiyem pri nizkikh temperaturakh)PERIODICAL: Fizika metallov i metallovedeniye, Vol 7, Nr 3, p 459
(USSR)

ABSTRACT: The rapid absorption of hydrogen by platinum black has enabled the authors to measure the isotherm of hydrogen absorption by palladium at -78°C . In Fig. 1 the results of three series of measurements are shown. The initial portion of the curve corresponds to the α -phase of the hydrogen-palladium solid solution; a horizontal portion of two-phase region follows, and the last portion rises steeply and corresponds to the β -phase of the solid solution. The pressure at which the phase transformation occurs is 0.015-0.018 mm Hg. The quantity of hydrogen dissolved at this pressure in the β -phase is 70-71 n. cm³/g Pd. At a pressure of 13 mm Hg the quantity of dissolved hydrogen is about 82 n. cm³/g Pd. (H/Pd = 0.78). At -120°C and a

Card 1/2 pressure of about 15 mm Hg the quantity of dissolved

67726

SOV/126-7-3-31/44

Absorption of Hydrogen by Palladium at Low Temperatures

hydrogen attained 86 n. cm³/g Pd (H₂/Pd = 0.82).
At -196°C absorption of hydrogen proceeded very
slowly. The experiments, carried out in a closed circuit
containing a given quantity of hydrogen and palladium, showed
that the hydrogen absorption still continued after two days.
The concentrations of hydrogen dissolved in palladium
remained lower than those at -78°C at the same pressures.
It follows that the hydrogen-palladium system had not reached
equilibrium as the solubility of hydrogen in palladium must
increase with drop in temperature.
There is 1 figure, and 4 references of which 2 are Soviet
and 2 English.

SUBMITTED: September 2, 1957

Card 2/2

SUTYRIN, A. S.

Dissertation: "The Organization of the Basic Production Process in the Flour Mill." Cand
Tech Sci, Moscow Technological Inst of the Food Industry, 28 Jun 54. (Vechnyaya Moskva,
Moscow 18 Jun 54)

SO: SUM 318, 23 Dec 1954

SUTYRIN, A.S.; BORINEVICH, A.A., redaktor

[Organizing and planning enterprises of the flour-milling industry]
Organizatsiia i planirovanie predpriatii mukomol'noi promyshlennosti.
Moskva, Izd-vo tekhn. i ekon. lit-ry po voprosam mukomol'no-krupianoi,
kombikormovoi promyshl. i elevatorno-skladskogo khoziaistva, 1956.
226 p. (MLRA 10:9)

(Flour-mills)

SUTYRIN, A.S. kand. tekhn. nauk.

Technical and organizational planning in flour and groats mills.
Trudy MTEPP no.7:58-65 '57. (MIRA 10:12)
(Grain milling)

SUTYRIN, B.V., voyenny letchik pervogo klassa, podpolkovnik; SERIKOV, B.S.,
mayor.

Competition for the outstanding crew and the best flight unit.
Vest.Vozd.Fl. 40 no.7:40-44 J1 '57. (MIRA 10:11)
(Russia--Air force)

SUTININ, G. N.

Dissertation: "Strength of Welds in the Process of Soling at Low Temperatures." Sand
Tech Sci, Moscow Higher Technical School imeni Bauman, Moscow, 1953. Referativnyy
Zhurnal--Khimiya, No 8, Apr 54.

SO: SUM 284, 26 Nov 1954.

SUPYRIN, G.V., kand.tekhn.nauk

High-quality welding with semiconductor units. Izv.vys.ucheb.zav.;
mashinostr. no.7:96-100 '60. (MIRA 13:11)

1. Moskovskoye vyssheye tekhnicheskoye uchilishche imeni Buamana.
(Electric welding)

SUTYRIN, G.V., kand.tekhn.nauk; NIKOL'SKAYA, Z.A., inzh.

Introduction of mechanized welding methods at the Kaluga
Machinery Manufacture. Svar.proizv. no.5:24-26 My '62.
(MIRA 15:12)

1. Kaluzhskiy vecherniy fakul'tet Moskovskogo vysshego
tekhnicheskogo uchilishcha im. Baumana (for Sutyryn).
2. Kaluzhskiy mashinostroitel'nyy zavod (for Nikol'skaya).

(Kaluga--Machinery industry)
(Electric welding--Equipment and supplies)

L 62541-65 EWP(k)/EWA(c)/EWT(d)/EWT(m)/EWP(i)/EWP(b)/T/EWA(d)/EWP(l)/EWP(v)/EWP(t)

JD/HM

UR/3155/64/000/001/0008/0015

ACCESSION NR: A15017526

AUTHOR: Sutyrin, G.V. (Docent, Cand. of technical sciences); Lobitskiy, V.G.
(Engineer)

25
87

TITLE: Unilateral spot welding of large equipment components

SOURCE: Moscow, Vysheye tekhnicheskoye uchilishche. Kaluzhskiy fakul'tet.
Mashinostroyeniya, no. 1, 1964, 8-15

TOPIC TAGS: unilateral spot welding, large equipment element, cumbersome machine part, welding setup, welding gun, spot gun, low alloy steel, low carbon steel, high alloy steel, copper busbar, welding transformer

ABSTRACT: To facilitate the welding of cumbersome and large machine parts, the authors employed the principle of unilateral spot welding. Fig. 1 shows the diagram of the setup for this process. The welding current from secondary winding 10 of transformer 5 is admitted to weldment 4 by copper cable 3 and to welding gun 1 by cable 2. The pressure between the electrode and the weldment is created by the welder by pressing the handle of gun 1 and thus compressing

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L 62541-65

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ACCESSION NR: AT5017526

the spring inside the gun's housing and causing the contacts 4 of the switch to close. Thereupon, contactor 6 energizes transformer 5 and timing relay 7, thus automatically causing the welding process to commence. For greater flexibility and lighter weight, current-conducting cable 2 consists of copper cores with an overall cross section of approx. 40 mm² sheathed in a rubber hose through which cooling water is passed. Numerous experiments have shown that the welding gun (Fig. 2) can be successfully used to weld large work parts of low-carbon, low-alloy and high-alloy steels in cases where, for one reason or another, the conventional machines for spot welding in any spatial position cannot be employed. Unilateral single-spot welding with the aid of the spot gun can find broad application in the welding of sheets up to 2 mm thick to elements of considerable thickness (as much as 600 mm and more), which cannot be accomplished with standard welding machines and pinon guns. This technique can be used to weld the standard welding machines and pinon guns. The length of the gun should not exceed 3 m.

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L 62541-65

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ACCESSION NR: AT5017526

and the attendant underwelding losses that may arise when the weldment is of large dimensions and the distance from the current source is considerable. The quality of the weldments thus obtained is sufficiently high. Orig. art. has: 2 figures, 3 tables.

ASSOCIATION: none

SUBMITTED: 00

NO REF SOV: 000

ENCL: 02

SUB CODE: IE, MM

OTHER: 000

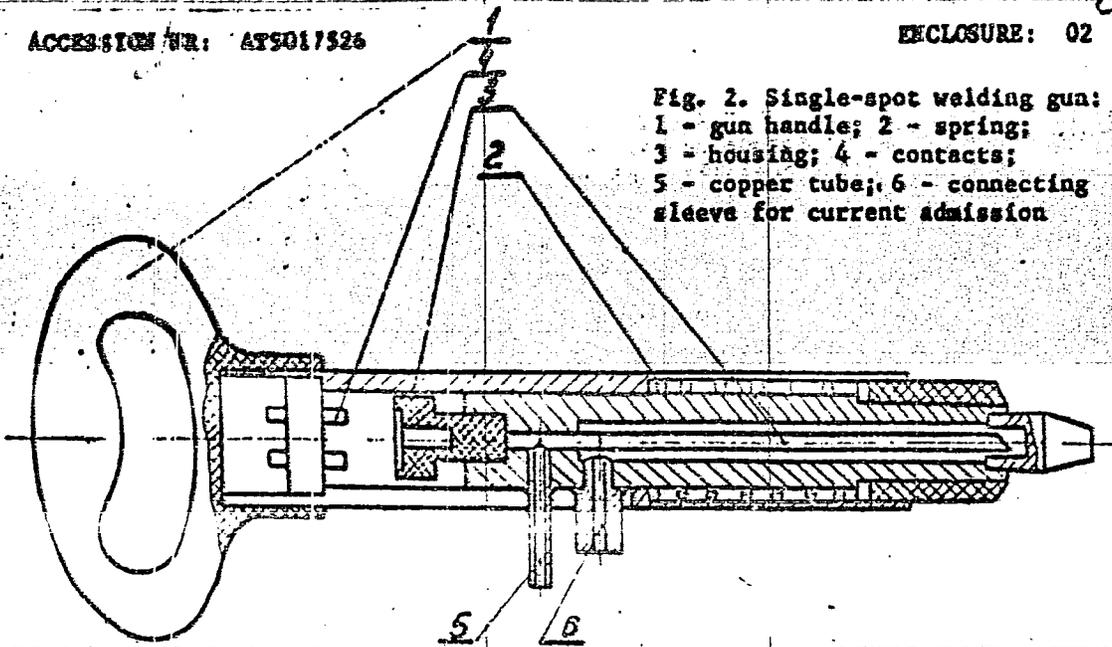
Card 3/5

L 62541-65

ACCESSION NR: AT5017526

ENCLOSURE: 02

Fig. 2. Single-spot welding gun:
1 - gun handle; 2 - spring;
3 - housing; 4 - contacts;
5 - copper tube; 6 - connecting
sleeve for current admission



Card

5/5

SUTYRIN, M., inzh.

Mechanizing the lowering of dried-out ships following salvage.
Rech. transp. 21 no.9:56 S '62. (MIRA 15:9)
(Dry docks) (Sluice gates)

SUTYRIN, M., inzh.

Do not omit the study of special sailing directions. Rech.
transp. 21 no.10:46-47 0 '62. (MIRA 15:10)

(Inland navigation)

SUTYRIN, M., inzh.

Voluntary inspectors of shipping. Rech.transp. 21 no.11:42-43
N '62. (MIRA 15:11)

(Shipping)

SUTYBIN, M. A.

The steering gear of river vessels. Kuibyshev. Rechizdat, 1942. 219 p. (49-33935)

VK145.88

SUTYRIN, M. A.

"Manual for River-Pilot", published by State Publishers of River Transport
Literature, Moscow, 1949

SMYRIN, M.A.

Prakticheskoe posobie rulevoru (shturval'novu). [Practical steering manual]. Izd. 2.
Utverzhdeno v kachestve ucheb. posobija dlia podgotovki rulevykh (shturval'nykh).
Moskva, Izd-vo Ministerstva rechnogo flota SSSR, 1950. 290 p. illus.

Bibliography: p. [238].

First ed. published in 1942 under title: Rulevoi
rechnogo sudna. DLC: V841.S8 1950

SO: Soviet Transportation and Communications, A Bibliography, Library of Congress,
Reference Department, Washington, 1952, Unclassified.

SUTYRIN, M. A.

Inland Navigation - Study and Teaching

Methods of training navigators.
Rech. Transp. 12, no. 1, 1952.

9. Monthly List of Russian Accessions, Library of Congress, June 1952. UNCLASSIFIED.

SUTYRIN, Mikhail Andreyevich; SEDOV, F.G., redaktor; MARKUSHINA, A.N.,
redaktor; BEGICHEVA, M.N., tekhnicheskiy redaktor

[Textbook for the training of helmsmen of river and lake vessels]
Uchebnik dlia podgotovki rulevogo rechnykh i ozernykh sudov. Mo-
skva, Izd-vo "Rechnoi transport," 1954. 242 p. (MLRA 8:6)
(Inland navigation)

SUTYRIN, M.A., inzhener.

More on draft changes during the movement of ships. Rech.transp.
14 no.4:23-24 Ap '56. (MLRA 9:8)
(Displacement (Ships))

SUTYRIN M.A.

PLCHATIN, A.A., inzh.; BEN'KO, M.P.; KAMENSKIY, V.K.; KARTASHEV, R.D.;
SUTYRIN, M.A.; FADMEYEV, V.G., red.; IGOSHIN, M.G., red.; KARYAKINA,
M.S., tekhn.red.

[Manual for helmsmen for lifesaving cutters] Posobie motoristu-
rulevomu spasatel'nogo katera. Moskva, 1957. 188 p. (MIRA 11:5)

1. Vsesoyuznoye dobrovol'noye obshchestvo sodeystviya armii,
aviatsii i flotu.
(Navigation) (Motorboats)

SUTYRIN, M.A.

Do not allow ship breakdowns in 1957. Rech. transp. 16 no.3:4-5
Mr '57. (MLRA 10:4)

(Inland navigation)

SUTYRIN, Mikhail Andreyevich; GLAZKOV, M.M., red.; KHARYUKOV, N.A., retsenzent;
ALEKSEYEV, V.I., red. izd-va; GORCHAKOV, G.N., tekhn. red.

[Safety in pushing barges with towboats] Bezopasnoe vozhdenie sudov
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